## Chromatographic approaches for determination of low-molecular mass aldehydes in bio-oil

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In the context of phenol/formaldehyde resin formulation, the aldehyde fraction of bio-oil is very important. Additionally, the toxicity of some volatile aldehydes, especially formaldehyde (Permissible exposure limit (PEL) in air at working places is 0.75 mg L<sup>-1</sup> [5]) demands their quantification in bio-oil. The presence of hydroxyacetaldehyde (glycolaldehyde), formaldehyde and acetaldehyde in bio-oil, has been previously described [2–4,6]. Determination of hydroxyacetaldehyde has been carried out using direct injection to a gas chromatograph coupled to a mass selective detector for identification of compounds and flame ionization detector coupled to the system (GC/MS/FID) for quantification [6]. Other aldehydes, as propionaldehyde and different substituted furaldehydes, can also be detected and quantified using this system [6].

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Are these the most representative references?

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Detailed characterization of the pyrolytic liquids obtained by pyrolysis of sawdust

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On the other hand, quantification of formaldehyde and acetaldehyde in bio-oil has not been deeply achieved yet. Determination of these aldehydes in bio-oil using high performance liquid chromatography (HPLC) preceded by derivatization with 2,4-dinitrophenylhydrazine (DNPH) has been mentioned in only one reference, without enough information about the analytical quality of the obtained results [7]. This approach has been widely described

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In a recent study, Sinag et al. [] used ion chromatography to analyze the bio-oil obtained from sawdust and reported that the contents of formaldehyde and acetaldehyde were dependent on the pyrolysis temperature.

Table 5

Determination of aldehyde-PFBHA oxime in bio-oil samples, by two optimized method.

Bio-oil samples	Formaldehyde (wt%)		Acetaldehyde (wt%)		Propionaldehyde (wt%)	
	D-HS	OFD-HS-SPME	D-HS	OFD-HS-SPME	D-HS	OFD-HS-SPME
1	1.6 ± 0.3	1.9 ± 0.2	$0.08 \pm 0.03$	0.03 ± 0.01	$0.03 \pm 0.02$	$0.012 \pm 0.007$
2	$0.8 \pm 0.2$	$1.3 \pm 0.3$	$0.03 \pm 0.01$	$0.01 \pm 0.01$	ND	LD>LQ
3	$1.8 \pm 0.3$	$2.2 \pm 0.2$	ND	$0.03 \pm 0.01$	ND	LD>LQ
4	$0.9 \pm 0.1$	$1.7 \pm 0.2$	$0.09 \pm 0.04$	$0.12 \pm 0.07$	ND	$0.04 \pm 0.01$
5	$1.1 \pm 0.3$	$1.7 \pm 0.2$	$0.07 \pm 0.01$	$0.04 \pm 0.01$	ND	$0.010 \pm 0.005$
6	$3.2 \pm 0.5$	$3.6 \pm 0.4$	ND	$0.05 \pm 0.01$	ND	$0.023 \pm 0.007$
7	$2.1 \pm 0.4$	$2.4 \pm 0.3$	$0.06 \pm 0.01$	$0.02 \pm 0.01$	ND	$0.014 \pm 0.003$
8	$2.3\pm0.4$	$2.0 \pm 0.2$	$0.11\pm0.04$	$0.03\pm0.01$	ND	$0.010 \pm 0.003$

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showing a p-value higher than 0.05. Finally, in Table 5 the quantitative results for aldehydes in different bio-oil samples are shown.

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It is interesting to note that the contents of formaldehyde and acetaldehyde (in these bio-oil samples derived from ??) are similar to the values reported by Sinag et al. [] (for sawdust-derived bio-oil), which ranged (as a function of pyrolysis temperature) from 0.0 to 2.6% and 0.04 to 0.17%, respectively.